Q.P. Code: 19HS0801

R19

Reg. No:

SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR

(AUTONOMOUS)

B.Tech I Year I Semester Supplementary Examinations December-2021 APPLIED CHEMISTRY

	(Common to EEE & ECE)		
T	ime: 3 hours	Max. Ma	arks: 60
	(Answer all Five Units $5 \times 12 = 60$ Marks)		
	UNIT-I		
1	Define Electrode Potential. Derive the Nernst equation for a single electrode	e L1	12M
	potential and write its applications.		
	OR		
2	a What is primary Battery? Write a brief note on Zinc-Air battery.	L1	6M
	b Write a short note on Alkali metal sulphide batteries.	L1	6M
UNIT-II			
3	a Explain Planck's Quantum Theory.	L2	6M
	b Write a brief note on particle in one dimensional box.	L1	6M
	OR		
4	Draw the molecular orbital diagrams of Oxygen molecule (O2) and Nitrogen molecule	L2	12M
	(N2). Explain their magnetic nature and bond order.		
	UNIT-III		
5	a Explain the Anionic addition polymerization mechanism with example.	L2	6M
	b Explain the Condensation or Step growth polymerization mechanism with example.	L2	6M
OR			
6	a Distinguish between Thermoplastics and thermosetting plastics.	L4	6M
	b Describe the preparation, properties and uses of Bakelite.	L2	6M
	UNIT-IV		
7 ′	Explain principle and instrumentation of UV-visible spectroscopy with neat diagram.	L2	12M
	OR		
8	What are the methods do you follow to separate from the Liquid Mixtures?	L1	12M
	UNIT-V		
9	Write a brief note on Fullerenes and Carbon nano tubes	L1	12M
OR			
10	a Write a note on Super Capacitors.	L1	6M
	b Define Dielectrics. What are the characteristics of Electrical Insulators?	L1	6M
	AAA DADAAA		